

In situ diode laser cavity ring-down measurements of acetylene in laminar flames

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Continuous-wave cavity ringdown spectroscopy





- Acetylene is involved in the hydrogen abstraction-acetylene addition (HACA) mechanism for soot growth.
- It is also involved in other important intermediate reactions during combustion.
- It is useful to obtain more data to validate soot models.
- Several other techniques have been used to measure acetylene.

Continuous-wave cavity ringdown spectroscopy





- Ring-down traces were averaged and recorded for each wavelength increment.
- Between each increment, we tuned to a reference wavelength at which absorption by acetylene is insignificant. This gives a baseline.

Continuous-wave cavity ringdown spectroscopy





The ring-down time of the cavity is given by:

$$\tau = \frac{L_c}{c[(1-R) + \alpha(\lambda)d + X]}$$

au is extracted from the decay signal by fitting it to an exponential function of the form: $I(t) = I_0 exp\left(-\frac{t}{\tau}\right)$

The absorption coefficient $\alpha(\lambda)$ is calculated from the ring-down time of the cavity and is given by: $\alpha(\lambda) = \frac{L_c}{cd} \left(\frac{1}{\tau(\lambda)} - \frac{1}{\tau_0}\right)$

Acetylene cw-CRDS spectra



Spectra for a range of different equivalence ratio flames recorded 2mm above the burner surface.

 Acetylene spectrum simulated from Hitran at a concentration of 1.22%, at 1600K, under atmospheric pressure.





Acetylene cw-CRDS spectra



Spectra for a range of different equivalence ratio flames recorded 2mm above the burner surface after background subtraction.



Concentration profiles



- Measured concentration of acetylene as a function of height above the burner surface.
- Previous temperature measurements by OH LIF



Concentration profiles



 Measured and calculated (ABF & USC mechanisms) concentrations of acetylene as a function of height above the burner surface.

Conclusion



- *In situ* measurements of acetylene have been obtained in an ethylene/air flame.
- Measurements show acetylene concentration increasing with increasing equivalence ratio.
- Acetylene concentration decreases with increasing height above the burner. This would be consistent with the consumption of acetylene including in reactions forming polycyclic aromatic hydrocarbons (PAHs) and ultimately soot.
- ABF and USC mechanisms predict a trend for acetylene concentration that rapidly increases and plateaus with increasing height above the burner surface.